



naming compounds containing polyatomic pdf

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Naming Compounds That Contain Polyatomic Ions Worksheet

2.1namingpolyatomicions.pdf - Writing Formulas for Compounds Containing Polyatomic ions When more than one ion is present in a formula unit, you must put parentheses around the polyatomic ion

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Naming Compounds Containing Polyatomic Ions Worksheet

Naming and Formula Writing for Ionic Compounds Containing Polyatomic Ions: Polyatomic ions are ions that consist of many atoms with an overall charge. For example, the polyatomic ion hydroxide consists of an oxygen bound to a hydrogen with an overall negative charge. The formula for hydroxide is OH^- . Another example, sulfate (SO_4^{2-}), consists of four oxygens bound to a sulfur with an overall charge of 2^- . Naming and formula writing for compounds containing polyatomic ions is very similar ...

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Naming Ionic Compounds With Polyatomic Ions Worksheet

Naming ionic compounds that contain polyatomic ions is very similar to naming binary ionic compounds. For example, the compound NaOH is called sodium hydroxide, because it contains the Na (sodium) cation and the OH (hydroxide) anion. To name these compounds, you must learn to recognize the common polyatomic ions.

Naming Compounds That Contain Polyatomic Ions

The properties of compounds containing polyatomic ions are very similar to those of binary ionic compounds.

The ions are arranged in a regular lattice and held together by coulombic forces of attraction.

6.17: Ionic Compounds Containing Polyatomic Ions

3. Name or write the formula for the following Type II polyatomic ionic compounds iron (III) bromate
 $\text{Ni}(\text{MnO}_3)_3$ copper (I) cyanate CrSO_5

Polyatomic Ions Worksheet

Part A: Naming ionic compounds that contain polyatomic ions Formula Name Table 1: NaBrO_4 sodium perbromate Na_2SO_4 sodium sulfate Na_3PO_4 sodium phosphate 1. Explain why the different compounds in Table 1 above have different numbers of Na atoms in their formulas. Your answer should discuss the charges on the various polyatomic ions in the three compounds. Formula Name Table 2: MgCrO_4 $\text{Al}(\text{C}_2\text{H}_3\text{O}_2) \dots$

Part A: Naming ionic compounds that contain polyatomic

Worksheet 5.1 Writing and Naming Ionic Compounds with Polyatomic Ions and Transition Metals Section A Write the name of the ionic compounds containing polyatomic ions

Answer Key Worksheet 5.1 Naming and Writing Ionic

How to name ionic compounds containing common polyatomic ions ... If a compound contains more than one polyatomic ion of the same type, ...

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Compounds containing polyatomic ions are named similarly to those containing only monatomic ions, except there is no need to change to an "ide" ending, since the suffix is already present in the name of the anion. Examples are shown in Table (PageIndex{2}).

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